## What is claimed is:

- 1 1. An organometallic compound having bonds between metal atoms
- 2 and nitrogen atoms or bonds between semimetal atoms and nitrogen
- 3 atoms, wherein, the chlorine content of 200 ppm or less and the
- 4 water content of 30 ppm or less.
- 1 2. An organometallic compound according to claim 1, wherein the
- 2 general formula of the compound is represented by the following
- 3 formula (1):

$$M[(R^{1})_{2}N]_{(n-s)}(R^{2})_{s}$$
 (1)

- 5 (wherein, M represents a metal atom or semimetal atom, with
- 6 the metal atom being Hf, Zr, Ta, Ti, Ce, Al, V, La, Nb or Ni,
- 7 and the semimetal atom being Si, R<sup>1</sup> represents a methyl group or
- 8 ethyl group, R<sup>2</sup> represents an ethyl group, n represents the
- 9 valence of M, and s represents an integer of 0 to n-1).
- 1 3. An organometallic compound according to claim 2, wherein when
- 2 M is Hf in formula (1), the chlorine content in the compound is
- 3 200 ppm or less and the water in the compound content is 30 ppm
- 4 or less.
- 1 4. An organometallic compound according to claim 2, wherein when
- 2 M is Si in formula (1), the chlorine content in the compound is
- 3 1 ppm or less and the water content in the compound is 30 ppm

- 4 or less.
- 1 5. A synthesis method of an organometallic compound comprising:
- 2 obtaining a crude product of an organometallic compound by
- 3 using a metal-containing compound and aminolithium;
- 4 distilling said crude product under reduced pressure in a
- 5 vacuum distillation step to obtain a purified product of said
- 6 organometallic compound; and,
- 7 remoing impurities contained in the purified product in an
- 8 impurity removal step using flash chromatography following said
- 9 vacuum distillation step.
- 1 6. A synthesis method according to claim 5, wherein said
- 2 impurity removal step comprises:
- 3 forming a filler layer inside a pressure-resistant column
- 4 by filling into said column a slurry prepared by adding a filler
- 5 to a developing solvent;
- 6 injecting said purified product into a top of the filler
- 7 layer; and,

1

- 8 passing said purified product through the filler layer by
- 9 supplying a pressurized gas at a predetermined flow rate into
- 10 the column from a top of the column to adsorb impurities contained
- 11 in said purified product in the filler layer.
  - 7. A synthesis method according to claim 5, wherein said

- 2 impurities removed from the purified product are chlorine and
- 3 water.
- 1 8. A synthesis method according to claim 6, wherein said
- 2 impurities removed from the purified product are chlorine and
- 3 water.
- 1 9. A synthesis method according to claim 6, wherein said
- 2 developing solvent is at least one organic solvent selected from
- 3 the group consisting of n-alkane, diethyl ether and
- 4 dichloromethane, and a water content in the organic solvent is
- 5 30 ppm or less.
- 1 10. A synthesis method according to claim 6, wherein said filler
- 2 in the column comprises at least one type of particle selected
- 3 from the group consisting of SiO<sub>2</sub> particles, Al<sub>2</sub>O<sub>3</sub> particles, ZrO<sub>2</sub>
- 4 particles, TiO<sub>2</sub> particles and HfO<sub>2</sub> particles having a mean
- 5 particle diameter of  $0.3-0.5 \mu m$ , and a particle size distribution
- 6 width  $d_{90}/d_{10}$  of 0.8-1.2.
- 1 11. A synthesis method according to claim 6, wherein said
- 2 pressure-resistant column is a glass column having a diameter
- 3 of 10-20 cm and a height of 30-50 cm.
- 1 12. A synthesis method according to claim 10, wherein said

- 2 pressure-resistant glass column is filled with 500-1000 g of
- 3 column filler.
- 1 13. A synthesis method according to claim 11, wherein said
- 2 pressure-resistant glass column is filled with 500-1000 g of
- 3 column filler.
- 1 14. A synthesis method according to claim 6, wherein the
- 2 pressurized gas is Ar gas, a pressure of the pressurized gas is
- 3 1-2 kg, and a column flow rate has a spatial velocity (SV value)
- 4 of 2-4 cm/min.
- 1 15. A synthesis method according to claim 5, wherein said
- 2 metal-containing compound is hafnium chloride, zirconium
- 3 chloride, tantalum chloride, titanium chloride, cerium chloride,
- 4 vanadium chloride, lanthanum chloride, niobium chloride, nickel
- 5 chloride or silane tetrachloride.
- 1 16. A synthesis method according to claim 5, wherein said
- 2 aminolithium is obtained by reacting one of dimethylamine and
- 3 diethylamine with n-butyllithium.
- 1 17. A synthesis method according to claim 5, wherein when said
- 2 metal-containing compound is hafnium chloride, the resulting
- 3 organometallic compound is one of tetraquis

- 4 (dimethylamino) hafnium and tetraquis (diethylamino) hafnium.
- 1 18. A synthesis method according to claim 15, wherein when said
- 2 metal-containing compound is hafnium chloride, the resulting
- 3 organometallic compound is one of tetraquis
- 4 (dimethylamino) hafnium and tetraquis (diethylamino) hafnium.
- 1 19. A synthesis method according to claim 5, wherein when the
- 2 metal-containing compound is silane tetrachloride, the resulting
- 3 organometallic compound is one of tetraquis
- 4 (dimethylamino) silane and tetraquis (diethylamino) silane.
- 1 20. A synthesis method according to claim 15, wherein when the
- 2 metal-containing compound is silane tetrachloride, the resulting
- 3 organometallic compound is one of tetraquis
- 4 (dimethylamino) silane and tetraquis (diethylamino) silane.
- 1 21. A solution raw material containing an organometallic
- 2 compound according to claim 1 dissolved in an organic solvent.
- 1 22. A solution raw material containing an organometallic
- 2 compound according to claim 2 dissolved in an organic solvent.
- 1 23. A solution raw material containing an organometallic
- 2 compound according to claim 3 dissolved in an organic solvent.

- 1 24. A solution raw material comprising an organometallic
- 2 compound according to claim 4 dissolved in an organic solvent.
- 1 25. A solution raw material comprising an organometallic
- 2 compound obtained by a synthesis method according to claim 5
- 3 dissolved in an organic solvent.
- 1 26. A solution raw material comprising an organometallic
- 2 compound obtained by a synthesis method according to claim 6
- 3 dissolved in an organic solvent.
- 1 27. A solution raw material according to claim 21, wherein said
- 2 organic solvent is at least one type of compound selected from
- 3 the group consisting of n-alkane, tetrahydrofuran, cyclohexane,
- 4 cycloalkane and branched alkane.
- 1 28. A solution raw material according to claim 22, wherein said
- 2 organic solvent is at least one type of compound selected from
- 3 the group consisting of n-alkane, tetrahydrofuran, cyclohexane,
- 4 cycloalkane and branched alkane.
- 1 29. A solution raw material according to claim 23, wherein said
- 2 organic solvent is at least one type of compound selected from
- 3 the group consisting of n-alkane, tetrahydrofuran, cyclohexane,
- 4 cycloalkane and branched alkane.

- 1 30. A solution raw material according to claim 24, wherein said
- 2 organic solvent is at least one type of compound selected from
- 3 the group consisting of n-alkane, tetrahydrofuran, cyclohexane,
- 4 cycloalkane and branched alkane.
- 1 31. A solution raw material according to claim 25, wherein said
- 2 organic solvent is at least one type of compound selected from
- 3 the group consisting of n-alkane, tetrahydrofuran, cyclohexane,
- 4 cycloalkane and branched alkane.
- 1 32. A solution raw material according to claim 26, wherein said
- 2 organic solvent is at least one type of compound selected from
- 3 the group consisting of n-alkane, tetrahydrofuran, cyclohexane,
- 4 cycloalkane and branched alkane.
- 1 33. A metal-containing thin film produced by metal organic
- 2 chemical vapor deposition using an organometallic compound
- 3 according to claim 1.
- 1 34. A metal-containing thin film produced by metal organic
- 2 chemical vapor deposition using an organometallic compound
- 3 according to claim 2.
- 1 35. A metal-containing thin film produced by metal organic
- 2 chemical vapor deposition using an organometallic compound

- 3 according to claim 3.
- 1 36. A metal-containing thin film produced by metal organic
- 2 chemical vapor deposition using an organometallic compound
- 3 obtained by a synthesis method according to claim 5.
- 1 37. A metal-containing thin film produced by metal organic
- 2 chemical vapor deposition using an organometallic compound
- 3 obtained by a synthesis method according to claim 6
- 1 38. A metal-containing thin film produced by metal organic
- 2 chemical vapor deposition using a solution raw material according
- 3 to claim 21.
- 1 39. A metal-containing thin film produced by metal organic
- 2 chemical vapor deposition using a solution raw material according
- 3 to claim 22.
- 1 40. A metal-containing thin film produced by metal organic
- 2 chemical vapor deposition using a solution raw material according
- 3 to claim 23.
- 1 41. A metal-containing thin film produced by metal organic
- 2 chemical vapor deposition using a solution raw material according
- 3 to claim 24.

- 1 42. A metal-containing thin film produced by metal organic
- 2 chemical vapor deposition using a solution raw material according
- 3 to claim 25.
- 1 43. A metal-containing thin film produced by metal organic
- 2 chemical vapor deposition using a solution raw material according
- 3 to claim 26.